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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/524,826	02/18/2005	Daniel Jansen	2001-1379	1882
466 YOUNG & TH	7590 06/18/200 OMPSON	EXAMINER		
209 Madison Street			MERKLING, MATTHEW J	
	Suite 500 ALEXANDRIA, VA 22314			PAPER NUMBER
			1795	
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			06/18/2009	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)			
	10/524,826	JANSEN ET AL.			
Office Action Summary	Examiner	Art Unit			
	MATTHEW J. MERKLING	1795			
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence address			
A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING DA - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be tim vill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).			
Status					
Responsive to communication(s) filed on 20 Ma This action is FINAL . 2b)☑ This Since this application is in condition for allowar closed in accordance with the practice under E	action is non-final. nce except for formal matters, pro				
Disposition of Claims					
4) ☐ Claim(s) 20-39 is/are pending in the application 4a) Of the above claim(s) 29-38 is/are withdraw 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 20-28 and 39 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/or Application Papers 9) ☐ The specification is objected to by the Examine 10) ☐ The drawing(s) filed on is/are: a) ☐ access	n from consideration. relection requirement. r. epted or b) objected to by the B				
Applicant may not request that any objection to the or Replacement drawing sheet(s) including the correction 11). The oath or declaration is objected to by the Ex.	on is required if the drawing(s) is obj	jected to. See 37 CFR 1.121(d).			
Priority under 35 U.S.C. § 119					
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received.					
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date 2/18/05.	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:	ate			

Application/Control Number: 10/524,826 Page 2

Art Unit: 1795

DETAILED ACTION

Election/Restrictions

1. Applicant's election without traverse of Group I (claims 20-29) in the reply filed on 3/20/09 is acknowledged.

Information Disclosure Statement

2. The information disclosure statement filed 2/18/05 fails to comply with 37 CFR 1.98(a)(2), which requires a legible copy of each cited foreign patent document; each non-patent literature publication or that portion which caused it to be listed; and all other information or that portion which caused it to be listed. It has been placed in the application file, but the information referred to therein has not been considered.

Claim Rejections - 35 USC § 103

- 3. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 4. The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
 - 1. Determining the scope and contents of the prior art.

Application/Control Number: 10/524,826 Page 3

Art Unit: 1795

2. Ascertaining the differences between the prior art and the claims at issue.

- 3. Resolving the level of ordinary skill in the pertinent art.
- 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 5. Claims 20 and 22-28 are rejected under 35 U.S.C. 103(a) as being unpatentable over Vollmar et al. (US 6,162,556) in view of Koga (US 6,033,634) and Deckman et al. (US 6,830,596).

Regarding claim 20, Vollmar discloses a method for converting CO and water into hydrogen (shift reaction, see abstract) wherein the feed (conduit 10) to the shift reactor (30) comprises anode off-gas (via conduit 10) from a fuel cell (4).

Vollmar teaches the use of a shift reactor on an anode off-gas stream and further discloses the use of a hydrogen separation apparatus (36) downstream of the shift reactor to separate the hydrogen from the carbon dioxide, but does not explicitly disclose the structure of the shift reactor. In other words, Vollmar fails to teach:

converting CO on one side of a membrane in the presence of water to CO2 and H2O on said one side of said membrane, H2 passing through said membrane to the other side of said membrane and said hydrogen being combusted on said other side with oxygen fed to said other side.

Koga also discloses a shift reactor used for converting CO and H2O into hydrogen and CO2 (see abstract).

Koga teaches a membrane shift reactor which receives feed on one side of the membrane (in channel 8, see Fig. 1) and converts the CO in the feed in the presence of CO2 and H2O into H2 and CO2 (shift reaction, see abstract). Koga then teaches permeating the produced hydrogen through a membrane (12, 13, 14) and providing a

Art Unit: 1795

stream of hydrogen (see Fig. 1). In other words, Koga teaches an integrated shift reactor and hydrogen separation method.

As such, it would have been obvious to one of ordinary skill in the art at the time of the invention to integrate the hydrogen separation step and the shift reaction step of Vollmar (as taught by Koga) in order to provide a more compact system (with the combination of two structures into one). Furthermore, the use of one piece of construction (as taught by Koga) instead of the multiple pieces of construction (as taught by Vollmer) would be merely a matter of obvious engineering choice and would have been obvious to one of ordinary skill in the art at the time of the invention (see MPEP 2144.04 (V)(B)).

Furthermore, Vollmar, as modified above by Koga, teaches a shift/membrane reactor which produces a stream of hydrogen gas which can be used for a plurality of purposes (see col. 6 lines 2-5 of Vollmar). However, Vollmar does not teach combusting the hydrogen on the permeate side of the membrane.

Deckman also discloses a membrane reactor which comprises a shift catalyst and separates the hydrogen from the reaction effluent by allowing only the hydrogen to permeate through the membrane (see abstract).

Deckman teaches combusting the hydrogen on the permeate side of the membrane in order to produce a gas which can be used to power a turbine and generate electricity (see abstract).

As such, it would have been obvious to one of ordinary skill in the art at the time of the invention to add the hydrogen combustion process step of Deckman, to the process of Art Unit: 1795

Vollmar, in order to produce a gas which can be used to power a turbine and generate electricity.

Regarding claim 22, Vollmar further discloses that non-combusted oxygen (from source 12) is fed to a cathode (22) of a downstream fuel cell (4).

Regarding claim 23, Vollmar further discloses that said oxygen comprises is from air (col. 6 lines 58-65).

Regarding claim 24, Vollmar, as modified above, further discloses water is separated off from the off-gas originating from said one side of said membrane (water is separated in 34).

Regarding claim 25, Vollmar further discloses the heat from the off-gas from at least one of the sides of said membrane is recovered (via heat exchanger 32, for example, see Fig. 1).

Regarding claim 26, Vollmar, as modified by Deckman above, further discloses an oxygen-containing gas (for combustion of the hydrogen, as mentioned above) is introduced on said other side of the membrane under elevated pressure (effluent is fed to a turbine, indicating that the feed to the turbine is at a substantially elevated pressure).

Regarding claims 27 and 28, Vollmar does not disclose gas containing water originating from the other side of said membrane is fed to a further step for converting CO on one side of a further membrane in the presence of water to give CO2 and H20 on the one side of said further membrane, H2 passing through said further membrane to the other side of said further membrane. However, such a modification is nothing more than a duplication of the membrane disclosed by modified Vollmar and discussed above.

Providing a duplicate membrane would amount to a mere duplication of parts and process steps. It has been held that mere duplication of parts has no patentable significance unless a new and unexpected result is produced. *In re Harza*, 274 F.2d 669, 124 USPQ 378 (CCPA 1960).

6. Claims 21 and 39 are rejected under 35 U.S.C. 103(a) as being unpatentable over Vollmar et al. (US 6,162,556) in view of Koga (US 6,033,634) and Deckman et al. (US 6,830,596) as applied to claim 20 above and further in view of Iio et al. (US 2002/0068204).

Regarding claims 21 and 39, Vollmar teaches a cathode off-gas which comprises air (col. 7 lines 35-43) but fails to teach that the oxygen used to combust the hydrogen comes from the cathode off-gas.

Iio also discloses a process in which the anode-off gas from a fuel cell is combusted to generate heat (see abstract).

Iio teaches combusting the hydrogen-containing anode off-gas with oxygen from the cathode off-gas as a means to supply oxygen to the combustion reaction (see paragraph 24).

As such, it would have been obvious to one of ordinary skill in the art at the time of the invention to combust the hydrogen of modified Vollmar, with the cathode exhaust gas (which comprises air), as taught by Iio as a preferable means to provide oxygen to the hydrogen combustion reaction.

Application/Control Number: 10/524,826 Page 7

Art Unit: 1795

Conclusion

7. Any inquiry concerning this communication or earlier communications from the

examiner should be directed to MATTHEW J. MERKLING whose telephone number is

(571)272-9813. The examiner can normally be reached on M-F 8:30-4:30.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Alexa Neckel can be reached on (571) 272-1446. The fax phone number for the

organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent

Application Information Retrieval (PAIR) system. Status information for published applications

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like assistance from a USPTO Customer Service Representative or access to the automated

information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/M. J. M./

Examiner, Art Unit 1795

/Alexa D. Neckel/

Supervisory Patent Examiner, Art Unit 1795